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## Table 1

The	pressure	coefficient	of	the	first	exciton	peak	in	the	optical	spectra	of	thallous.
				lea	id, a	nd bism	uth ha	lid	les				

	E o (eV)	$(\partial E_0 / \partial p)_T$ (10 <sup>-6</sup> eV/bar)	т ( <sup>о</sup> к)	$(\partial E/\partial T)$ average $(10^{-4} \text{ eV/deg})$
TlBr cubic	3.0	-20.1 + 1.0 -9.5 + 0.4	274 80	+3.4
PbI <sub>2</sub> layer	2.5	-18.5 + 1.0 -16.5 + 0.5	294 80	-1.25
BiI <sub>3</sub> layer	2.0	-16.6 + 1.0 -11.2 + 0.8	274 80	-2.6

perimental investigations (8) and energy band calculations (9) for TlBr have established that the  $E_0$  exciton peak in TlBr is associated with a valence band maximum in which there is a significant contribution from the Tl ion 6s-states. The appropriate conduction band is formed from Tl p-states. The large negative pressure coefficients which are obtained in the case of PbI<sub>2</sub> and BiI<sub>3</sub> therefore provide strong evidence of significant contributions from metal 6s-states to the upper valence band in each material, since states of s-like symmetry rise in energy much faster under pressure than do p- or d-like states.

The detailed band structure of  $PbI_2$  and  $BiI_3$  is not known, and an unequivocal assignation of the transition associated with the peak  $E_0$  in each material to a specific critical point in the Brillouin zone cannot be made. However, considerations of band curvature suggest that in common with the thallous halides, the first transition in both PbI<sub>2</sub> and BiI<sub>3</sub> is likely to occur at the zone boundary.

It was found for thallous bromide (8) that the positive temperature coefficient of the  $E_0$  peak energy may be attributed substantially to the effect of lattice dilatation, with a relatively small electron-lattice interaction term  $(\partial E/\partial T)_V$ . In the case of PbI<sub>2</sub> and BiI<sub>3</sub> however, the energy of the first peak in optical absorption has both a negative temperature coefficient (10, 7) and a large negative pressure coefficient, so that the electron-lattice interaction term must be negative in both materials, with a magnitude greater than that of the total temperature coefficient.

K30